Simple 1,2-Diamine Ligands for Asymmetric Addition of Aryllithium Reagents to Imines

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Enantioselective addition of various aryllithium reagents to aromatic imines was catalyzed (20 mol-%) by readily accessible 1,2-diamines to afford a wide range of protected diarylmethylamines in up to 94% enantiomeric excess. Further-

more, the absolute configuration of these arylation products was determined by using X-ray crystallography. (© Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, Germany, 2005)

Introduction and Background

Chiral amines are compounds of great interest in organic synthesis since they constitute building blocks for complex structures of biologically active molecules, as well as being employed as chiral auxiliaries, ligands, or agents for kinetic resolution.[1] The addition of aliphatic organometallic compounds to the azomethine bond of activated imines is now a route of choice for synthesizing aliphatic amines and arylalkylamines in excellent yields and enantioselectivities.^[2] By contrast, the asymmetric addition of aromatic compounds to imines affording diarylmethylamines has been scarcely reported, although this backbone can be found in different drugs such as cetirizine or the non-peptidic γ -opioid agonist SNC-80.^[3] In 1990 Tomioka and coworkers described the first enantioselective addition of an aryllithium compound to aromatic imines promoted by an external chiral ligand, but despite this work, [4] only a few efficient examples of such arylation processes were reported in the nineties. Nevertheless, since 2000, this is a topic of growing interest. The most remarkable examples in terms of enantioselectivity include the diphenylzinc addition to N-formylimines reported by Bräse et al. in 2002^[5] and the work of Hayashi et al.^[6] They described in 2000 a highly enantioselective rhodiumcatalyzed arylation of N-alkylidenesulfonamides with organostannanes, and very recently a more environmentally friendly procedure involving the addition of arylboroxines to different imines with enantioselectivities of up to 99%.[6] All of these procedures are indeed high yielding and selective, but they suffer from the cost of the catalysts used. We report herein a low cost and efficient catalytic arylation of

Results and Discussion

Recently, we described the synthesis of new chiral tertiary 1,2-diamines 1a-f (Scheme 1) and their use in the addition of organolithium compounds to N-p-methoxyphenylimines.^[8] Herein, we wish to present the results obtained for the addition of aryllithium reagents to these types of imines (Table 1). Bearing in mind that only a few examples have been reported to date on this topic, our preliminary study and optimization of the conditions were carried out with imine 2, the substrate used by Tomioka and co-workers in 1994.^[9] They reported that in the presence of a stoichiometric amount of diether 4, product 3 was obtained in 87% ee with the absolute configuration $S^{[9]}$. We found that the same reaction afforded 3 in 86% ee (Entry 7). As also mentioned by these authors, we noticed that only toluene and diethyl ether were suitable solvents for the enantioselective version of the reaction. Indeed, in more coordinating solvents such as THF, the reaction proceeded well, but afforded a racemic product. When the reaction was carried out with our diamine ligands, poorer levels of enantioselectivity were induced, particularly with hindered ligands, in contrast with the results obtained for the previously reported addition of methyllithium to imines.[8] It has to be noted at this point that the best result (54% ee, Entry 1) was reached with the simplest (R,R)-(-)-N,N,N',N'-tetramethyl-trans-1,2-cyclohexanediamine (1a), readily prepared by the Eschweiler-Clark methylation of the readily available (R,R)-1,2-diammoniumcyclohexane mono-(+)-tartrate.[10]

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aromatic imines using a wide diversity of aryllithium reagents activated by readily accessible chiral 1,2-diamines.^[7] The stereochemical outcome of the reaction has been established by a single X-ray analysis of an *N-p*-methoxyphenyl-diarylmethylamine adduct [(-)-16, see text].

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Scheme 1.

Table 1. Addition of phenyllithium to imine 2.

Entry	Ligand	Isolated yield (%)	ee (%)[a]
1	1a	72	54 (R)
2	1b	60	19 (R)
3	1c	61	0
4	1d	34	11 (R)
5	1e	44	43 (R)
6	1f	79	19 (R)
7	4	49	86 (S) ^[b]
8	(–)-sparteine	50	23 (R)

[a] Enantiomeric excesses were determined by supercritical fluid chromatography (SFC), column Chiralcel OB-H. The absolute configuration was established by comparison of the SFC chromatogram with the one obtained in entry 7 (see ref.^[9]). [b] 87% *ee* was reported by Tomioka et al. (see ref.^[9]).

When employing aromatic imines 5a and 5b as substrates (Scheme 2),^[11] with the expectation that this system is more suitable than imine 2 for our diamine ligands, a different trend was observed. We found that under stoichiometric conditions, and at -78 °C, an enantiomeric excess of 58% was obtained with the addition of phenyllithium to imine 5a with diamine 1a (Table 2, Entry 1), versus 44% with (-)-sparteine (Entry 7) and only 7% with diether 4 (Entry 6). Moreover, under substoichiometric conditions, the level of enantioselectivity remained unchanged with 0.2 equiv. of ligand 1a per mol of imine (Entry 2), whereas such catalysis was not possible with (-)-sparteine (Entry 8). It should be pointed out that, in the toluene solvent, the reaction is rather sluggish in the absence of diamine. For example, the synthesis of the racemic adducts was performed at room temperature, otherwise the conversions were very low. Good induction was also obtained with ligands 7a [prepared from readily available (1S,2S)-N-methylpseudoephedrine following the procedure of Dieter][12] and 7b[13] (61% and 59% ee, respectively, Entries 9 and 10). In both cases, the major enantiomer was the opposite one to that obtained with ligand 1a. The effect of the configuration of each stereocenter of the 1,2-diamine ligand on the enantioselectivity was pointed out by using 7c [prepared from (1R,2S)-N-methylephedrine following the procedure of Dieter^[12] as the catalyst (Entry 11).

Scheme 2.

Table 2. Addition of phenyllithium to imines 5a and 5b.

Entry	Ligand (0.2 equiv.)	Product	Isolated yield (%)	ee (%) ^[a]
1	1a (2 equiv.)	6a	69	58 (+)
2	1a	6a	95	60 (+)
3	1a	6b	87	20 (-)
4	1c	6a	68	3
5	1e	6a	62	12
6	4 (2 equiv.)	6a	62	7 (+)
7	(–)-sparteine (2 equiv.)	6a	60	44 (-)
8	(–)-sparteine	6a	79	0
9	7a	6a	82	61 (-)
10	7b	6a	87	59 (–)
11	7c	6a	81	28 (+)

[a] Enantiomeric excesses were determined by SFC, columns Chiralcel OJ for **6a** and Chiralcel AD-H for **6b**. Signs in parenthesis refer to the optical rotations.

The influence of an α -heteroatom could clearly be seen from the comparison of thienyl imine **5a** and furyl imine **5b**. Thienyl imine **5a** behaves like other aryl imines (see Table 3), with 60% *ee* (Table 2, Entry 2), whereas furyl imine **5b** affords a low 20% *ee* (Table 2, Entry 3). Thus, in contrast to oxygen, sulfur does not seem to compete with the diamine catalyst for coordination to lithium. The low *ee* may be ascribed to this oxygen–lithium coordination.

Table 3 shows the generalization of the method. With aromatic imines 9-11 (Scheme 3, Table 3) we could also evaluate the effect of the electronic character of the substrate. These results reveal that the electronic effect of the aromatic part of the imines 9–11 is low. Good yields were obtained in all cases with enantioselectivities between 60-70%. Thus, a typical electron-withdrawing group, such as trifluoromethyl (imine 9d, Entry 9) afforded an ee of 60%, whereas a typical electron-donating group, such as methoxy (imine **9b**, Entry 7) gave a higher *ee* of 68%. Only product 14 presented a slightly lower ee of 55% (Entry 14), perhaps due to a steric hindrance of the 1-naphthyl group. Once again, an efficient catalysis with ligand 1a was possible, while it was not with (-)-sparteine (compare Entries 2 and 4). The efficiency of the catalysis was tested further with only 10 and 5 mol-% of the diamine ligand. Despite the slightly lower conversions, the enantioselectivities remained almost unchanged (Entries 9–11).

Table 3. Addition of phenyllithium to imines 9–11.^[a]

Entry	Imine	Ligand	Product	% Conversion ^[b] (% isolated yield)	ee (%) ^[c]
1	9a	1a (3 equiv.)	12a	>99 (64)	64 (+)
2	9a	1a (0.2 equiv.)	12a	62 (42)	62 (+)
3	9a	(-)-sparteine (3 equiv.)	12a	88 (72)	59 (–) ^[d]
4	9a	(–)-sparteine (0.2 equiv.)	12a	39 (23)	2
5	9a	4 (3 equiv.)	12a	>99 (72)	28 (+)
6	9b	1a (3 equiv.)	12b	79 (78)	69 (–)
7	9b	1a (0.2 equiv.)	12b	26 (14)	68 (–)
8	9c	1a (0.2 equiv.)	12c	58 (42)	63 (–)
9	9d	1a (0.2 equiv.)	12d	>99 (72)	60 (+)
10	9d	1a (0.1 equiv.)	12d	97 (90)	56 (+)
11	9d	1a (0.05 equiv.)	12d	93 (86)	53 (+)
12	9e	1a (0.2 equiv.)	12e	48 (44)	64 (–)
13 ^[e]	10	1a (0.2 equiv.)	13	77 (74)	61 (–)
14	11	1a (0.2 equiv.)	14	79 (61)	55 (+)

[a] All reactions were carried out at -78 °C in toluene as the solvent for 4 h, unless otherwise stated. [b] Conversions evaluated by ¹H NMR spectroscopy of the crude. [c] Enantiomeric excesses were determined by SFC, columns Chiralpak AS-H for **12a** and **12d**, Chiralcel OJ for **12b** and Chiralcel OD-H for **12c**, **12e**, **13** and **14**. Signs in parenthesis refer to the optical rotations. [d] Reported in 50% *ee* under stoichiometric conditions by Senanayake (see ref.^[3b]). [e] Reaction time: 2.5 h.

$$\begin{array}{c} \text{PhLi (3 equiv.),} \\ \text{ligand} \\ \\ \text{PhLi (3 equiv.),} \\ \text{ligand} \\ \\ \text{Ph} \\ \text{N} \\ \text{N} \\ \text{Ph} \\ \text{OMe} \\ \\ \text{PoMe} \\ \\ \text{PoMe} \\ \\ \text{PoMe} \\ \text{PoMe} \\ \\ \text{PoMe} \\$$

Scheme 3.

One of the advantages of this approach is that by simply changing the imine/organolithium partner, either enantiomer can be produced. Indeed amines **6a** and **6b** could be obtained by addition of thienyl and furyllithium to imine **8**, formed in situ by deprotonation of furane and thiophene with *n*BuLi (Scheme 4). Amines **12a**, **12b**, **13** and **14** were formed by addition of the corresponding aryllithium reagents, obtained by metal–halogen exchange of the appropriate aryl halides and *n*BuLi (for details see Exp. Sect.). In

general, the enantioselectivities induced by ligand 1a were similar to those described above, despite the higher temperatures needed in some cases, to obtain reasonable yields (Table 4). Interestingly, the *ee* of 14 reached an exceptionally good 90% *ee* with the addition of 1-naphthyllithiun to imine 8 (Entry 6).

6a, 6b, 12a-b, 13, 14

Scheme 4.

Table 4. Addition of aryllithium compounds to imine 8.

Entry	Conditions	Ar in ArLi	Product	Isolated yield (%)	ee (%)[a,b]
1	1a (2 equiv.), toluene, –20 °C, 15 h	2-Thienyl	6a	71	58 (-) [58 (+)]
2	1a (2 equiv.), Et ₂ O, –78 °C to room temp.	2-Furyl	6b	45	28 (+) [20 (-)]
3	1a (2 equiv.), toluene, –78 °C, 4 h	4-ClC ₆ H ₄	12a	76	69 (–) [64 (+)]
4	1a (2 equiv.), toluene, –78 °C, 2 h	$4-\text{MeOC}_6\text{H}_4$	12b	93	62 (+) [69 (-)]
5	1a (2 equiv.), toluene, 78 °C, 2 h	2-naphthyl	13	98	67 (+) [61 (–)]
6	1a (2 equiv.), toluene, –78 °C, 2 h	1-naphthyl	14	98	90 (–) [55 (+)]

[a] Enantiomeric excesses were determined by SFC Columns Chiralcel OJ for 6a and 12b, Chiralcel AD-H for 6b, Chiralpak AS-H for 12a and Chiralcel OD-H for 13 and 14. Signs in parenthesis refer to the optical rotations. [b] The enantiomeric excesses of the addition of phenyllithium to imines, from Table 2 and Table 3 are indicated in brackets.

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In order to study the interesting case of 1-naphthyllithium more thoroughly, the addition was carried out on other imines (Scheme 5, Table 5), as done previously for phenyllithium. However, a dramatic difference of reactivity between these two aryllithium compounds was observed. While the addition of phenyllithium to imine 9a was complete in 4 h at -78 °C (Table 3, Entry 1), a conversion of only 15% after 6 h was reached for the addition of 1-naphthyllithium, but with a very good ee of 89% (Table 5, Entry 7). Similarly, with the less reactive imine 9e, at -78 °C after 7 h, no reaction was detected. Different temperatures were tested, and good yields were obtained at temperatures near to -40 or -50 °C without any significant loss of enantioselectivity (Entries 13-15). Moreover, when (-)-sparteine was tested as the ligand, poor enantioselectivities and yields were obtained (compare Entries 10 and 8). Ligand 7a afforded compound 14 in a 94% ee, one of the highest enantioselectivities reported for this kind of reaction (Entry 3). As we stated in the introduction, the achievement of a substoichiometric version of this promising asymmetric reaction would be the key to new perspectives and future

Scheme 5.

improvements. Thus we tested the addition of 1-naphthyllithium to imine 8 in the presence of 0.2 equiv. of 1a or 7a and were gratified to obtain respectively, 84% and 80% ee (Entries 2 and 4). These good results with substoichiometric amounts of ligands that are so easy to prepare are providing a future for this procedure in the arylation of imines.

Since the interchange of aryl imines and aryllithiums produces opposite enantiomers, it is clear that any aryllithium attacks the same prochiral face of any aryl imine. To determine this facial selectivity, it was necessary to know the absolute stereochemistry of our adducts. Thus, the absolute configuration of the major enantiomer of (-)-16 could be determined by X-ray crystallography. Indeed, this amine was obtained in 89% ee by addition of 1-naphthyllithium to imine 8 in Et₂O as solvent and in the presence of 2 equiv. of 1a (Entry 9). After usual purification by column chromatography, the amine was obtained as a white solid and crystals suitable for X-ray analysis were obtained from a mixture of pentane/Et₂O. We determined the configuration to be R [Flack parameter x = -0.01 (9)].^[14] CD spectra of both (-)-16 and the crystal analyzed by X-ray crystallography were of the same sign, therefore confirming the Rconfiguration of (-)-16. 1-naphthyllithium activated by 1a thus attacks the Si face of the imine. It seems reasonable to assume that the same stereochemical pathway is followed by all the other aryllithium reagents. Although the Si face is attacked, the actual absolute configuration may vary according to the CIP priority rules.

Conclusions

In summary, a wide variety of N-protected diarylmethylamines, in both enantiomeric forms, were synthesized with enantiomeric excesses of up to 94% by addition of aryl-

Table 5. Addition of 1-naphthyllithium to different imines.

Entry	R	Imine	Product	Ligand	Conditions ^[a]	% Conversion ^[b] (% isolated yield)	ee (%) ^[c]
1 ^[d]	C ₆ H ₅	8	14	1a	−78 °C, 2 h	>99 (98)	90 (-)
2	C_6H_5	8	14	1a (0.2 equiv.)	−78 °C, 4 h	19	84 (–) ^[e]
3	C_6H_5	8	14	7a	−78 °C, 2 h	70 (45)	94 (+)
4	C_6H_5	8	14	7a (0.2 equiv.)	−78 °C, 9 h	57	80 (+)
5	$2\text{-MeC}_6\text{H}_4$	9f	15	1a	−41 °C, 3 d ^[f]	>99	86 (-) (>98) ^[g]
5	$2-MeC_6H_4$	9f	15	7a	−25 °C, 5 d ^[f]	>99	94 (+)
7	4-ClC ₆ H ₄	9a	16	1a	−78 °C, 6 h	15	89 (-)
3	4-ClC ₆ H ₄	9a	16	1a	−65 °C, 21 h	>99 (64)	90 (-)
)	4-ClC ₆ H ₄	9a	16	1a	−65 °C, 21 h ^[h]	67	89 $(-)$ $(R)^{[i]}$
10	4-ClC ₆ H ₄	9a	16	(–)-sparteine	−67 °C, 13 h	50	10
11	$4-MeC_6H_4$	9c	17	1a	−55 °C, 3 d ^[f]	>99 (66)	88 (-)
12	4-MeC ₆ H ₄	9c	17	7a	−55 °C, 3 d ^[f]	28 (28)	92 (+)
13	$3,4-(OCH_2O)-C_6H_3$	9e	18	1a	−55 °C, 3 d ^[f]	20	90 (–)
4	$3,4-(OCH_2O)-C_6H_3$	9e	18	1a	−41 °C, 3 d ^[f]	>99 (60)	86 (-)
15	$3,4-(OCH_2O)-C_6H_3$	9e	18	7a	−25 °C, 4 d ^[f]	>99 (90)	90 (+)

[a] Unless otherwise indicated, all reactions were carried out in toluene and in the presence of 2 equiv. of the ligand. [b] Conversions evaluated by 1H NMR spectroscopy of the crude. [c] Enantiomeric excesses were determined by SFC Columns Chiralcel OD-H for 14 and 18 and OJ for 15–17. Signs in parenthesis refer to the optical rotations. [d] From Table 4. [e] Average of two runs. [f] Nonoptimized reaction times. [g] In parenthesis, enantiomeric excess after recrystallization in Pentane/Et₂O. [h] Experiment carried out in Et₂O. [i] Determined by X-ray analysis.

lithium compounds to aromatic imines catalyzed by readily available 1,2-diamine ligands. The absolute configuration of one of these diarylmethylamines could be established by single-crystal X-ray analysis. Further studies on the mechanism are underway in our laboratory.

Experimental Section

General Techniques: Diethyl ether and toluene were dried over Na/ benzophenone and CaH₂, respectively, and then distilled under nitrogen. TMEDA was distilled prior to use and stored under argon. Other commercially available reagents were used without further purification, unless otherwise indicated. Imines 2, [9] 5, [15] 8, [9] 9a, [16] 9b,^[17] 9c,^[18] 9d,^[19] 9e,^[20] 9f, 10,^[21] and 11,^[22] were generated by condensation of p-anisidine with the corresponding aldehyde. Ligands 1b-f,[8] 7a,[12] 7c[12] and 7b[13] were synthesized as described in the literature. nBuLi was purchased as a 1.6 m solution in hexanes and PhLi as a 1.9 m solution in cyclohexane/Et₂O, 70:30. All reactions were carried out under inert argon and in oven-dried glassware. Temperatures indicated in the text refer to the inner reaction mixtures. Flash column chromatography was performed using silica gel 32-63 μm, 60 Å and thin layer chromatography on silica gel plates. A 254 nm UV light and an ethanolic solution of ninhydrine are used as developing agents. ¹H NMR and ¹³C NMR spectra were recorded at room temperature at 400 and 100 MHz, respectively. Chemical shift values are expressed in ppm relative to residual CHCl₃ calibrated to 7.26 for ¹H NMR spectroscopy and the CDCl₃ signal is calibrated to 77.0 for ¹³C NMR spectroscopy. Enantiomeric excesses were determined by Supercritical Fluid Chromatography (capillary column, 10 psi H₂), using racemic mixtures as references. Temperature programs are described as follows: initial temperature (°C); initial time (min); temperature gradient (°C·min⁻¹); final temperature (°C). Circular dichroism spectra were recorded in acetonitrile as solvent. The measurements were carried out in the far-UV region.

Synthesis of (R,R)-(-)-N,N,N',N'-Tetramethyl-trans-1,2-cyclohexanediamine (1a): (R,R)-1,2-Diammonium cyclohexane mono-(+)-tartrate^[10] (24 g, 0.091 mol) was dissolved in formic acid 85% (36 mL) and formal dehyde 40% (44 mL) was added slowly at room temperature. The mixture was refluxed for 2 h. After cooling to room temperature, and then to 0 °C, the reaction mixture was made basic until pH 12 with solid sodium hydroxide and extracted thoroughly with diethyl ether. The combined organic layers were dried over Na₂SO₄, filtered, and concentrated under reduced pressure. The residue was purified by distillation in a bulb-to-bulb apparatus (b.p. 50 °C/0.1 Torr) to afford a colorless liquid (11.35 g, 74%) of $[\alpha]_{10}^{20} = -62.9$ (c = 1.05, CHCl₃). Spectral data were in agreement with the literature. [23]

Synthesis of (1*S*,2*S*)-*N*,*N*,*N'*,*N'*-Tetramethyl-1-phenylpropane-1,2-diamine (7a): Following the procedure of Dieter,^[12] mesyl chloride (5.59 mL, 72.0 mmol) was added dropwise under inert conditions to a solution of (+)-(1*S*,2*S*)-*N*-methylpseudoephedrine (5.42 g, 30.0 mmol) and triethylamine (12.5 mL, 90.0 mmol) in diethylether (180 mL) at 0 °C. A white precipitate appeared and after 30 min triethylamine (16.7 mL, 120 mmol) was added once again and the temperature was raised to room temperature. Dimethylamine (96 mL, 5.6 m in EtOH) was added and the solution was stirred vigorously overnight. The solution was made basic with aqueous NaOH and organic layer was extracted with diethyl ether, washed with 5% aq. NaHCO₃ and water. The residue was dried over Na₂SO₄, filtered and concentrated under reduced pressure. The

product was isolated in 75% yield as a yellow oil of $[\alpha]_D^{20} = -31.3$ (c = 1.01, CHCl₃) and was used without further purification. Spectral data were in agreement with the literature.^[24]

Synthesis of (1R,2S)-N,N,N',N'-Tetramethyl-1-phenylpropane-1,2-diamine (7e): The procedure was the same as described previously for 7a, starting from (-)-(1R,2S)-N-methylephedrine. The product was isolated in 85% yield as a white solid of $[\alpha]_D^{20} = -1.5$ (c = 0.95, CHCl₃) after recrystallization in diethylether. Spectral data were in agreement with the literature.^[24]

General Procedure for the Asymmetric Addition of Phenyllithium to Imines: A solution of imine (0.5 mmol) in dry toluene (9 mL) was added to a stirred solution of ligand (0.1 mmol or 1.5 mmol) in dry toluene (1 mL) at room temperature. After cooling the medium to -78 °C, phenyllithium (0.8 mL as a 1.9 м solution in cyclohexane/ Et₂O, 70:30) was slowly added under argon. The resulting mixture was stirred at -78 °C for 2 or 4 h (see Tables) and then quenched at low temperature with methanol and water. Layers separated and the aqueous layer was extracted three times with Et2O and once with EtOAc. The combined organic phases were dried over K₂CO₃, filtered and finally concentrated under reduced pressure. The resulting residue was analyzed by ¹H NMR spectroscopy to determine the conversion of the reaction by integration of the residual peak of the aldimino proton. Then, it was purified by silica gel column chromatography to give the desired product in the ee indicated in tables. Spectral data were in agreement with the corresponding racemic mixture synthesized independently.

General Procedure for the Asymmetric Addition of 2-Furyllithium, 2-Thienyllithium, 1-Naphthyllithium, 2-Naphthyllithium, 4-Chlorophenyllithium and 4-Methoxyphenyllithium to Imines: n-Butyllithium (0.6 mL as a 1.6 m solution in hexanes, 1 mmol) was slowly added under argon to a stirred solution of ligand (0.1 mmol or 1.0 mmol) and the appropriate aromatic compound (1.1 mmol) in dry toluene (10 mL) cooled to -60 °C. The resulting mixture was stirred at -60 °C for 1 h, during which time the metal-halogen exchange occurred. Then, the temperature of the medium was adjusted to the one indicated in the tables and a solution of imine (0.5 mmol) in toluene (10 mL) was added carefully, so that the temperature of the medium was only raised minimally. The reaction mixture was stirred for the time indicated in the text. Quenching and workup were the same as described previously for the addition of phenyllithium, as well as the determination of conversions. Flash silica gel column chromatography afforded the pure enantioenriched compounds, whose spectroscopic data were in agreement with the corresponding racemic mixture synthesized independently.

4-Methoxy-*N***-[phenyl(thiophen-2-yl)methyl]benzenamine (6a):** Purification by silica gel chromatography (toluene) gave (–)-**6a** as a yellow oil of $[a]_D^{25} = -13.1$ (c = 1.03, CHCl₃ for ee = 58% with **7b**); ee was determined by SFC: Chiralcel OJ, 200 bar, 2 mL·min⁻¹, 25% MeOH in CO₂ (25%, 10 min), 30 °C, $t_1 = 7.7$ min, $t_2 = 8.6$ min. ¹H NMR: $\delta = 7.50$ –7.45 (m, 2 H), 7.39 (t, J = 7.3 Hz, 2 H), 7.35–7.29 (m, 1 H), 7.25 (dd, J = 5.0, 1.3 Hz, 1 H), 6.99–6.96 (m, 1 H), 6.94–6.91 (m, 1 H), 6.76 (d, J = 8.8 Hz, 2 H), 6.60 (d, J = 8.8 Hz, 2 H), 5.72 (s, 1 H), 4.35–4.05 (br. s, 1 H), 3.74 (s, 3 H) ppm. ¹³C NMR: $\delta = 152.5$, 148.0, 142.8, 141.1, 128.8, 127.7, 127.1, 126.9, 125.0, 124.9, 115.0, 114.8, 59.6, 55.7 ppm. IR (neat): 3392, 3062, 1601, 1508, 1452, 1231, 1035, 819, 761, 698 cm⁻¹. MS (EI), m/z (%): 295 (M, 10), 174, 173 (100), 129 (10). HRMS calcd. for C₁₈H₁₇NOS 295.1031, found 295.1027.

N-[(Furan-2-yl)(phenyl)methyl]-4-methoxybenzenamine (6b): Purification by silica gel chromatography (toluene) gave (+)-6b as a yellow oil of $[a]_{25}^{25}$ = +5.9 (c = 0.50, CHCl₃ for ee = 28% with 1a); ee was determined by SFC: Chiralcel AD-H, 200 bar, 2 mL·min⁻¹, 5%

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MeOH in CO₂ (5%, 5 min, 1%·min⁻¹, 20%, 5 min), 30 °C, t_1 = 6.2 min, t_2 = 7.7 min. ¹H NMR: δ = 7.51–7.47 (m, 2 H), 7.45–7.39 (m, 3 H), 7.38–7.34 (m, 1 H), 6.80 (d, J = 9.1 Hz, 2 H), 6.63 (d, J = 8.8 Hz, 2 H), 6.77 (dd, J = 3.2, 1.9 Hz, 1 H), 6.18 (br d, J = 3.3 Hz, 1 H), 5.59 (s, 1 H), 4.40–4.00 (br. s, 1 H), 3.76 (s, 3 H) ppm. ¹³C NMR: δ = 155.4, 152.3, 142.0, 141.0, 140.6, 128.6, 127.6, 127.2, 114.8, 114.6, 110.2, 107.3, 57.6, 55.5 ppm. IR (neat): 3401, 2929, 1511, 1453, 1235, 1036, 819, 737, 699 cm⁻¹. MS (EI), m/z (%): 279 (M, 11), 172 (12), 158 (14), 157 (100), 128 (22), 127 (10), 105 (14), 95 (10), 77 (14). HRMS calcd. for C₁₈H₁₇NO₂ 279.1259, found 279.1239.

N-[(4-Chlorophenyl)(phenyl)methyl]-4-methoxybenzenamine (12a): Purification by silica gel chromatography (toluene) gave 12a as a yellow oil of $[\alpha]_{D}^{25} = -13.6$ [c = 0.97, CHCl₃ for ee = 59% with (–)-sparteine], +12.3 (c = 0.38, CHCl₃ for ee = 64% with 1a); ee was determined by SFC: Chiralpak AS-H, 200 bar, 2 mL·min⁻¹, 25% MeOH in CO₂ (25%, 15 min), 30 °C, $t_1 = 3.3$ min, $t_2 = 3.7$ min. ¹H NMR: $\delta = 7.24$ –7.16 (m, 9 H), 6.70–6.58 (m, 2 H), 6.47–6.32 (m, 2 H), 5.30 (s, 1 H), 3.88 (br. s, 1 H), 3.63 (s, 3 H) ppm. ¹³C NMR: $\delta = 152.4$, 142.8, 141.6, 141.4, 133.0, 128.9, 128.7, 127.6, 127.5, 114.8, 114.7, 63.3, 55.8 ppm. IR (neat): 3402, 3028, 1602, 1509, 1452, 1241, 1035, 819, 750, 700 cm⁻¹. MS (EI), mlz (%): 324 (M + H, 14), 323 (M, 64), 321 (82), 292 (M, 4), 244 (21), 216 (53), 201 (93), 165 (47), 139 (84), 111 (40), 105 (100), 77 (94), 51 (35). HRMS calcd. for C₂₀H₁₈³⁵CINO 323.1077, found 323.1059 and for C₂₀H₁₈³⁷CINO 325.1047, found 325.1071.

4-Methoxy-N-[(4-methoxyphenyl)(phenyl)methyl]benzenamine (12b): Purification by silica gel chromatography (pentane/Et₂O: 8:1.5) gave (-)-**12b** as a yellow oil of $[\alpha]_D^{25} = -23.2$ (c = 0.77, CHCl₃ for ee = 69% with **1a**); ee was determined by SFC: Chiralcel OJ, 200 bar, 2 mL·min⁻¹, 2% MeOH in CO₂ (2%, 15 min), 30 °C, $t_1 = 11.3$ min, $t_2 = 13.2$ min. 1 H NMR: $\delta = 7.29-7.16$ (m, 7 H), 6.80–6.75 (m, 2 H), 6.65–6.61 (m, 2 H), 6.42 (d, J = 8.0 Hz, 2 H), 5.29 (s, 1 H), 3.88 (br. s, 1 H), 3.70 (s, 3 H), 3.63 (s, 3 H) ppm. 13 C NMR: $\delta = 158.8$, 152.1, 143.4, 141.8, 135.5, 128.7, 128.6, 127.3, 127.2, 114.7, 114.1, 63.2, 55.8, 55.3 ppm. IR (neat): 3398, 3000, 1610, 1509, 1452, 1243, 1034, 819, 761, 700 cm⁻¹. MS (EI), m/z (%): 319 (M, 10), 197 (100), 165 (9), 153 (9). HRMS calcd. for $C_{21}H_{21}NO_2$ 319.1572, found 319.1580.

4-Methoxy-*N*-[phenyl(*p*-tolyl)methyl]benzenamine (12c): Purification by silica gel chromatography (pentane/Et₂O, 10:1) gave (–)-12c as a yellow oil of $[\alpha]_{25}^{25} = -17.3$ (c = 0.96, CHCl₃ for ee = 63% with 1a); ee was determined by SFC: Chiralcel OD-H, 200 bar, 2 mL·min⁻¹, 2% MeOH in CO₂ (2%, 5 min, 1%·min⁻¹, 15%), 30 °C, $t_1 = 13.0$ min, $t_2 = 13.5$ min. ¹H NMR: $\delta = 7.30-7.15$ (m, 7 H), 7.04 (d, J = 7.8 Hz, 2 H), 6.63 (d, J = 8.8 Hz, 2 H), 6.42 (d, J = 9.0 Hz, 2 H), 5.30 (s, 1 H), 3.90 (br. s, 1 H), 3.63 (s, 3 H), 2.24 (s, 3 H) ppm. ¹³C NMR: $\delta = 152.1$, 143.3, 141.7, 140.3, 136.9, 129.4, 128.7, 127.3, 127.2, 114.7, 63.6, 55.7, 21.1 ppm. IR (neat): 3403, 3026, 1602, 1509, 1452, 1242, 1037, 819, 732, 700 cm⁻¹. MS (EI), m/z (%): 304 (M + H, 5), 303 (M, 26), 182 (14), 181 (100), 166 (20), 165 (18), 152 (6), 77 (5). HRMS calcd. for C₂₁H₂₁NO 303.1623, found 303.1602.

4-Methoxy-*N*-{**[4-(trifluoromethyl)phenyl](phenyl)methyl}benzenamine (12d):** Purification by silica gel chromatography (pentane/ Et₂O, 10:1) gave (+)-**12d** as a yellow oil of $[\alpha]_D^{25} = +33.7$ (c = 1.01, CHCl₃ for ee = 60% with **1a**); ee was determined by SFC: Chiralpak AS-H, 200 bar, 2 mL·min⁻¹, 5% MeOH in CO₂ (5%, 6 min, 2%·min⁻¹, 15%), 30 °C, $t_1 = 3.0$ min, $t_2 = 4.0$ min. ¹H NMR: $\delta = 7.48$ (d, J = 8.4 Hz, 2 H), 7.42 (d, J = 8.4 Hz, 2 H), 7.24–7.12 (m, 5 H), 6.63 (d, J = 9.1 Hz, 2 H), 6.39 (d, J = 8.8 Hz, 2 H), 5.36 (s, 1 H), 3.91 (br. s, 1 H), 3.61 (s, 3 H) ppm. ¹³C NMR: $\delta = 152.5$,

147.1, 142.5, 141.2, 129.5 (q, ${}^2J_{\rm C,F} = 32.6$ Hz), 129.0, 127.8, 127.6, 127.5, 125.7 (q, ${}^3J_{\rm C,F} = 3.9$ Hz), 124.2 (q, ${}^1J_{\rm C,F} = 272.0$ Hz), 114.8, 63.6, 55.7 ppm. IR (neat): 3397, 3030, 1618, 1509, 1416, 1322, 1242, 1109, 1065, 1035, 817, 740, 700 cm $^{-1}$. MS (EI), m/z (%): 358 (M + H, 16), 357 (M, 67), 250 (15), 236 (18), 235 (87), 215 (9), 173 (10), 166 (17), 165 (32), 145 (9), 123 (11), 122 (100), 105 (26), 95 (10), 77 (13). HRMS calcd. for $C_{21}H_{18}F_3NO$ 357.1340, found 357.1358.

N-[(Benzo[*d*][1,3]dioxol-6-yl)(phenyl)methyl]-4-methoxybenzenamine (12e): Purification by silica gel chromatography (toluene) gave (–)-12e as a yellow oil of [α] $_{\rm D}^{25}$ –1.7 (c = 0.99, CHCl $_{3}$ for ee = 64% with 1a); ee was determined by SFC: Chiralcel OD-H, 175 bar, 1.8 mL·min $_{-}^{-1}$, 5% MeOH in CO $_{2}$ (5%, 5 min, 1%·min $_{-}^{-1}$, 20%), 30 °C, t_{1} = 15.8 min, t_{2} = 16.2 min. $_{-}^{1}$ H NMR: δ = 7.30–7.22 (m, 3 H), 7.19 (br. s, 2 H), 6.90–6.72 (m, 2 H), 6.70–6.58 (m, 3 H), 6.43 (d, J = 8.8 Hz, 2 H), 5.85 (s, 2 H), 5.25 (s, 1 H), 3.90 (br. s, 1 H), 3.64 (s, 3 H) ppm. $_{-}^{13}$ C NMR: δ = 152.2, 147.9, 146.7, 143.3, 141.6, 137.3, 128.7, 127.3, 127.2, 120.5, 114.7, 114.6, 108.3, 107.8, 101.0, 63.6, 55.7 ppm. IR (neat): 3397, 3027, 1610, 1509, 1231, 1035, 931, 818, 761, 700 cm $_{-}^{-1}$. MS (EI), mlz (%): 333 (M, 11), 331 (19), 255 (37), 254 (11), 240 (35), 226 (63), 212 (15), 211 (70), 153 (12), 152 (15), 150 (10), 149 (100), 121 (11), 105 (31), 77 (53), 63 (14), 51 (13). HRMS calcd. for C $_{21}$ H $_{19}$ NO $_{3}$ 333.1365, found 333.1379.

4-Methoxy-*N***-[(naphthalen-2-yl)(phenyl)methyl]benzenamine (13):** Purification by silica gel chromatography (toluene) gave (–)-**13** as a yellow oil of $[\alpha]_D^{25} = -1.6$ (c = 1.01, CHCl₃ for ee = 64% with **1a**); ee was determined by SFC: Chiralcel OD-H, 200 bar, 2 mL·min⁻¹, 25% MeOH in CO₂ (25%, 15 min), 30 °C, $t_1 = 10.4$ min, $t_2 = 11.5$ min. ¹HNMR (400 MHz, CDCl₃): $\delta = 7.85$ –7.72 (m, 3 H), 7.50–7.20 (m, 9 H), 6.71 (d, J = 8.8 Hz, 2 H), 6.57 (d, J = 8.8 Hz, 2 H), 5.58 (s, 1 H), 3.71 (s, 3 H), 1.55 (br. s, 1 H) ppm. ¹³CNMR (100 MHz, CDCl₃): $\delta = 134.5$, 133.4, 132.8, 130.1, 129.5, 129.1, 128.8, 128.5, 128.0, 127.6, 127.4, 126.1, 125.9, 125.8, 125.6, 114.7, 55.7 ppm. MS (EI), mlz (%) 339 (M, 18), 218 (15), 217 (100), 216 (11), 215 (21), 212 (11), 202 (23). HRMS calcd. for C₂₄H₂₁NO 339.1623, found 339.1646.

4-Methoxy-*N*-[(naphthalen-1-yl)(phenyl)methyl]benzenamine^[22] (14): Purification by silica gel chromatography (pentane/Et₂O, 10:1) gave 14 as a colorless oil of $[\alpha]_D^{25} = -40.7$ (c = 3.57, CHCl₃ for ee = 90% with 1a), + 60.1 (c = 0.32, CHCl₃ for ee = 94% with 7a); ee was determined by SFC: Chiralcel OD-H, 175 bar, 1.8 mL·min⁻¹, 20% MeOH in CO₂ (20%, 15 min), 30 °C, $t_1 = 9.6$ min, $t_2 = 12.2$ min.

4-Methoxy-*N***-[(naphthalen-1-yl)(***o***-tolyl)methyl]benzenamine (15):** Purification by silica gel chromatography (toluene) gave **15** as a white solid. M.p. 86–87 °C (pentane/diethyl ether). [α]₂⁵⁵ = -22.3 (c = 0.18, CHCl₃ for ee = 86% with **1a**), +36.6 (c = 0.40, CHCl₃ for ee = 94% with **7a**); ee was determined by SFC: Chiralcel OJ, 170 bar, 2 mL·min⁻¹, 10% MeOH in CO₂ (10%, 5 min, 2%·min⁻¹, 25%, 5 min), 30 °C, t_1 = 10.0 min, t_2 = 10.9 min. ¹H NMR (400 MHz, CDCl₃): δ = 7.95–7.75 (m, 4 H), 7.55–7.10 (m, 7 H), 6.73 (d, J = 8.8 Hz, 2 H), 6.49 (d, J = 7.4 Hz, 2 H), 6.25 (s, 1 H), 3.73 (s, 3 H), 2.31 (s, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 136.4, 134.3, 131.6, 131.0, 129.1, 128.4, 127.8, 127.6, 126.7, 126.5, 125.8, 125.8, 125.7, 123.6, 115.0, 55.9, 19.4 ppm. IR (neat): 3400, 2928, 1750, 1510, 1242, 1038, 819, 791, 758 cm⁻¹. MS (EI), m/z (%): 353 (29), 232 (22), 231 (100), 216 (28), 215 (30). HRMS: ESI (positive) calcd. for [M – H]+ C₂₅H₂₂NO 352.1701, found 352.1713.

N-[(4-Chlorophenyl)(naphthalen-1-yl)methyl]-4-methoxybenzenamine (16): Purification by silica gel chromatography (pentane/ Et₂O, 20:1) gave (-)-16 as a colorless oil of $[\alpha]_D^{25} = -36.9$ (c = 0.38, CHCl₃ for ee = 90% with 1a); ee was determined by SFC: Chiralcel OJ, 170 bar, 2 mL·min⁻¹, 10% MeOH in CO₂ (10%, 5 min, 2%·min⁻¹, 25%, 10 min), 30 °C, $t_1 = 23.0$ min, $t_2 = 24.7$ min. ¹H

NMR: δ = 8.04 (d, J = 7.3 Hz, 1 H), 7.94 (d, J = 7.6 Hz, 1 H), 7.84 (d, J = 7.8 Hz, 1 H), 7.55–7.22 (m, 8 H), 6.76 (d, J = 8.6 Hz, 2 H), 6.53 (d, J = 8.1 Hz, 2 H), 6.16 (s. 1 H), 4.07 (br. s, 1 H), 3.76 (s, 3 H) ppm. 13 C NMR: δ = 152.3, 141.4, 141.0, 137.6, 134.1, 133.2, 131.1, 129.4, 129.0, 128.9, 128.4, 126.6, 125.8, 125.7, 125.5, 123.4, 114.9, 114.3, 59.5, 55.7 ppm. IR (neat): 3403, 3051, 1597, 1508, 1240, 1014, 781, 735, 699 cm $^{-1}$. MS (EI), mlz (%): 374 (11), 373 (46), 372 (34), 371 (63), 370 (55), 260 (20), 253 (34), 252 (18), 251 (100), 249 (14), 217 (14), 216 (70), 215 (73), 127 (14), 92 (15), 77 (24), 64 (10). HRMS calcd. for $C_{24}H_{20}^{35}$ CINO 373.1233, found 373.1234.

4-Methoxy-*N*-[(naphthalen-1-yl)(*p*-tolyl)methyl]benzenamine (17): Purification by silica gel chromatography (toluene) gave 17 as a yellow oil of $[\alpha]_D^{25} = -86.6$ (c = 0.30, CHCl₃ for ee = 88% with **1a**), +47.9 (c = 0.14, CHCl₃ for ee = 92% with **7a**); ee was determined by SFC: Chiralcel OJ, 170 bar, 2 mL·min⁻¹, 10% MeOH in CO₂ $(10\%, 5 \text{ min}, 2\% \cdot \text{min}^{-1}, 25\%, 10 \text{ min}), 30 \,^{\circ}\text{C}, t_1 = 14.5 \,^{\circ}\text{min}, t_2 = 14.5 \,^{\circ}\text{min}$ 16.0 min. ¹H NMR: $\delta = 7.96$ (d, J = 7.1 Hz, 1 H), 7.80 (d, J =7.1 Hz, 1 H), 7.70 (d, J = 8.3 Hz, 1 H), 7.48–7.30 (m, 4 H), 7.28– 7.20 (m, 3 H), 7.12–7.03 (m, 1 H), 6.63 (d, J = 6.8 Hz, 2 H), 6.40 (d, J = 8.8 Hz, 2 H), 6.04 (s, 1 H), 3.97 (br. s, 1 H), 3.62 (s, 3 H),2.25 (s, 3 H) ppm. ¹³C NMR: δ = 152.0, 141.7, 139.5, 137.9, 137.1, 134.1, 131.1, 129.4, 128.9, 128.0, 126.3, 125.5, 125.3, 123.6, 114.8, 114.2, 59.8, 55.7, 21.1 ppm. IR (neat): 3400, 3000, 1510, 1242, 1030, 800, 785 cm⁻¹. MS (EI), m/z (%): 353 (M, 18), 232 (26), 231 (100), 230 (22), 229 (17), 217 (11), 216 (33), 215 (44), 123 (17), 108 (24), 80 (8). HRMS calcd. for C₂₅H₂₃NO 353.1780, found 353.1773.

N-{[Benzo]*d*][1,3]dioxol-6-yl](naphthalen-1-yl)methyl}-4-methoxybenzenamine (18): Purification by silica gel chromatography (toluene) gave 18 as a yellow oil of $[\alpha]_{25}^{15} = -59.1$ (c = 0.80, CHCl₃ for ee = 86% with 1a), +35.0 (c = 0.85, CHCl₃ for ee = 90% with 7a); ee was determined by SFC: Chiralcel OD-H, 170 bar, 2 mL·min⁻¹, 20% MeOH in CO₂ (20%, 30 min), 30 °C, $t_1 = 15.3$ min, $t_2 = 18.1$ min. 1 H NMR: $\delta = 8.10-8.00$ (m, 1 H), 8.00–7.75 (m, 2 H), 7.58–7.40 (m, 5 H), 6.98–6.85 (m, 2 H), 6.80–6.67 (m, 3 H), 6.53 (d, J = 8.6 Hz, 1 H), 6.07 (s, 1 H), 6.00–5.90 (m, 2 H), 3.72 (s, 3 H), 1.55 (br. s, 1 H) ppm. 13 C NMR: $\delta = 148.2$, 147.1, 134.3, 131.3, 129.1, 128.4, 126.6, 125.8, 125.7, 125.6, 123.7, 121.6, 115.0, 108.7, 108.6, 101.3, 55.9 ppm. IR (neat): 3400, 3000, 1750, 1511, 1237, 1038, 800 cm⁻¹. MS (EI), m/z (%): 383 (M, 12), 373 (46), 262 (25), 261 (100), 231 (29), 203 (34), 202 (32), 201 (8), 77 (4), 65 (4). HRMS calcd. for C₂₅H₂₁NO₃ 383.1521, found 383.1516.

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- [8] Other protecting groups such as N-benzyl and N-sulfonyl were not suitable for this reaction. For the synthesis of diamine ligands: a) J.-C. Kizirian, J.-C. Caille, A. Alexakis, *Tetrahedron Lett.* 2003, 44, 8893–8895. For the contribution of our laboratory to the addition of aryllithium compounds to aromatic imines: b) N. Cabello, J.-C. Kizirian, A. Alexakis, *Tetrahedron Lett.* 2004, 45, 4639–4642.
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